

# Transition Path Theory

Winterschool on  
Theoretical Chemistry and Spectroscopy  
9-13 December 2013, Han-sur-Lesse, Belgium

Bernd Ensing  
Computation Chemistry  
University of Amsterdam



Friday, 13 December, 13

## Summary

part 1

A stochastic hopping between states is Markovian if the probability to reach the next state does not depend on the past (no memory)

- reversible Markov chains (detailed balance)
- ergodicity
- spectral decomposition (convergence to equilibrium)

part 2

Markov State Modeling applied as analysis tool of the dynamics of a small peptide

- Analysis of molecular dynamics simulation
- Clustering based on structures
- Relation to free energy landscape based on cluster size
- Transition probability between clusters (states)
- Using the MSM the dynamics can be sampled in a coarse-grain manner, e.g. using kinetic Monte-Carlo

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# Content

- Lecture 1: Markov processes
- Lecture 2: State clustering
- Lecture 3: Transition Path Theory
- Lecture 4: Sampling Reactive Events

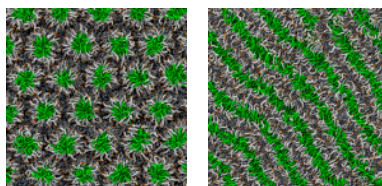
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## Molecular Transitions are Rare Events

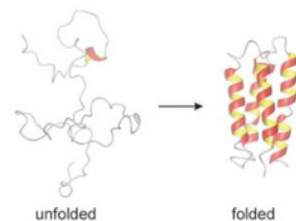
chemical reactions



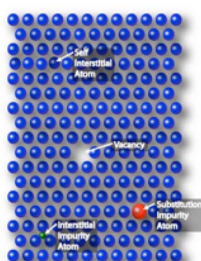
phase transitions



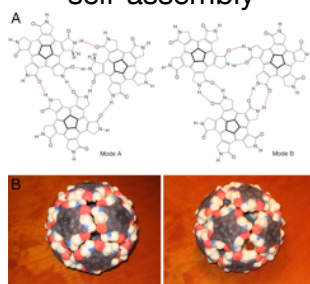
protein folding



defect diffusion



self-assembly



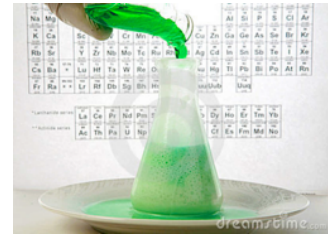
stock exchange crash



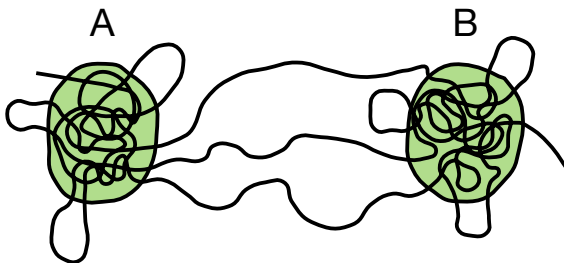
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# Rare Events

- Molecular transitions are not rare in every day life
- They are only rare events with respect to the femto-second timescale of atomic motions
- Modeling activated transitions by straightforward simulation would be extremely costly (takes forever)
- Advanced methods: Transition state theory (TST), Bennet-Chandler (Reactive Flux) approach, Transition Path Sampling (TPS), Free energy methods, Parallel Tempering, String Method,....



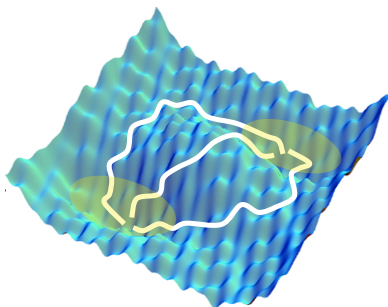
*A very long trajectory*



- A and B are (meta-)stable states, i.e. attractive basins
- transitions between A and B are rare
- transitions can happen fast
- system loses memory in A and B

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*Free energy landscape*



P.G. Bolhuis, D. Chandler,  
C. Dellago, P.L. Geissler  
*Annu. Rev. Phys. Chem 2002*

# Rare Events

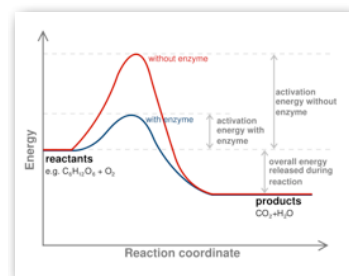
The free energy has local minima separated by barriers

Many possible transition paths via meta-stable states

Projection on reaction coordinate shows FE profile with transition state barrier

Reaction rate depends exponentially on the barrier height

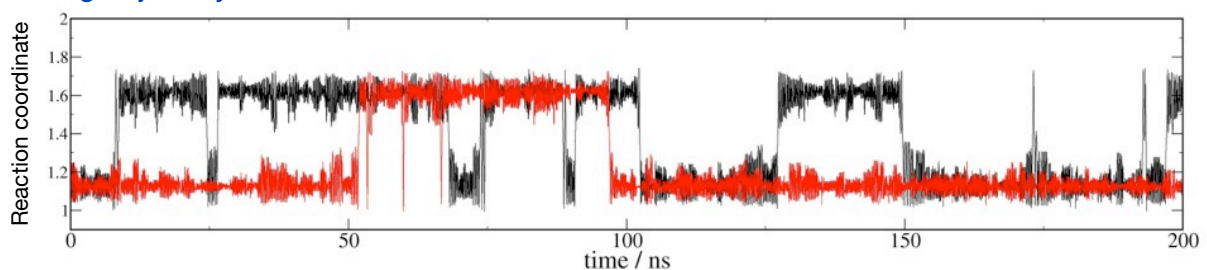
*Transition State*



$$k = k_0 e^{-\Delta G/KT}$$

Arrhenius equation

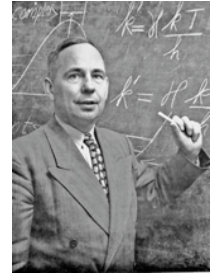
*A long trajectory*



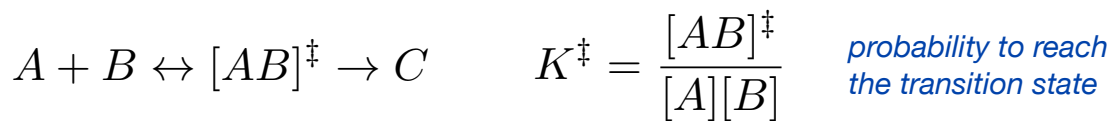
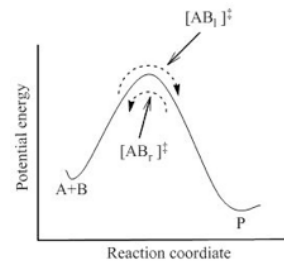
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# Transition State Theory

- TST  $\neq$  TPT
- Derived by Henry Eyring and by Meredith Gwynne Evans and Michael Polanyi in 1935
- Explains Arrhenius equation (1889)  $k = Ae^{-E_a/RT}$
- Activated complex is in quasi equilibrium with reactants
- All reactants that reach the activated complex state, continue to products



Henry Eyring



$$k = k^\ddagger K^\ddagger = \kappa \frac{k_B T}{h} e^{-\Delta G^\ddagger / RT}$$

←
←
  
frequency

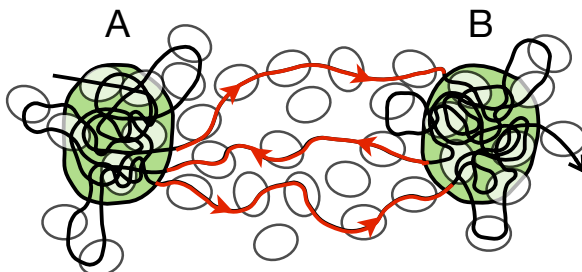
- rate is frequency of unstable mode times the TS probability
- transmission factor is  $\kappa=1$  in TST (correction for barrier recrossings)

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# Transition Path Theory

Analysis of reactive events to

- understand reaction mechanisms
- obtain accurate reaction rates



Eric Vanden-Eijnden

Transition paths are Markov chains through a space of intermediate states that depart from A and reach B before A

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# Some concepts

Committer function:  $q_i$  is the probability that a trajectory leaving from micro-state  $i$  arrives in B before A



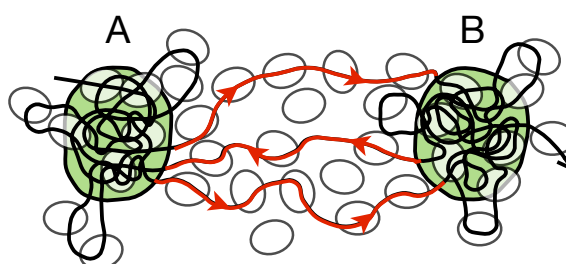
Lars Onsager:  
"Splitting  
probability"

The committer of a micro-state (or configuration) is computed by initializing a large number of trajectories from this state with random (Boltzmann distributed) velocities and recording how many arrive in B.

Micro-states close to A can be expected to have  $0 < q_i < < 1$  and close to B have  $0 < < q_i < 1$

$$q_i^B = 1 - q_i^A$$

In a sense, the committer function is the perfect reaction coordinate (although expensive and without mechanistic insight).



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# Some concepts



Lars Onsager:  
"Splitting  
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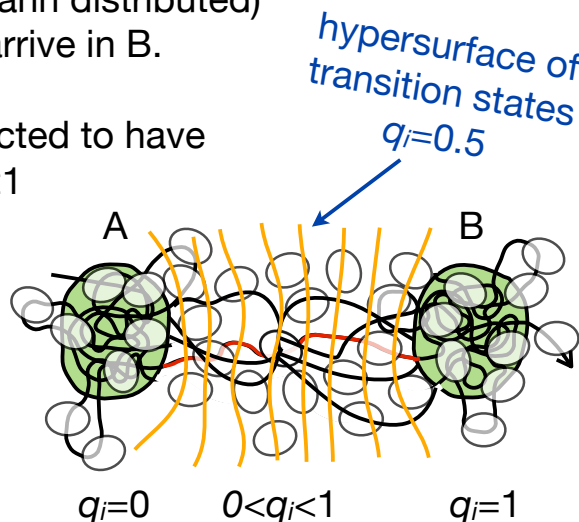
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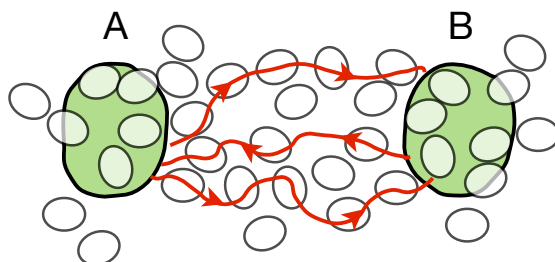
# Some concepts

Committer  $q_i = 0.5$  connects all micro-states that have a 50/50 chance to go either way. Perfect definition for transition states!

- there is not just 1 transition state at finite temperature
- no (free) energy surface saddle points involved
- accounts for dynamics and recrossings

$$q_i = \sum_{j \in B} t_{ij} + \sum_{j \notin A \cup B} t_{ij} q_j = \sum_{j=1}^N t_{ij} q_j$$

the committer is easily related to the transition probabilities to reach the next micro-state



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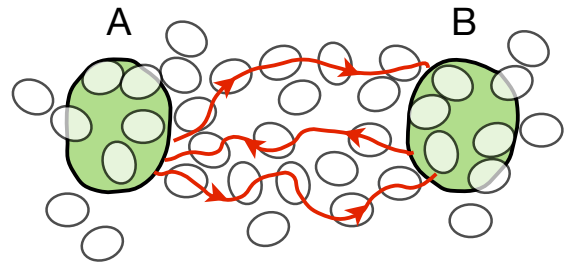
# Some concepts

Probability to observe a trajectory in  $i$   
and that it is reactive:

$$\pi_i^R = (1 - q_i)\rho_i q_i \quad \text{peaks in neighborhood of } q_i=0.5$$

Probability that the trajectory visits  $i$   
and  $j$  consecutively and it is reactive:

$$\pi_{ij}^R = (1 - q_i)\rho_i t_{ij} q_j$$



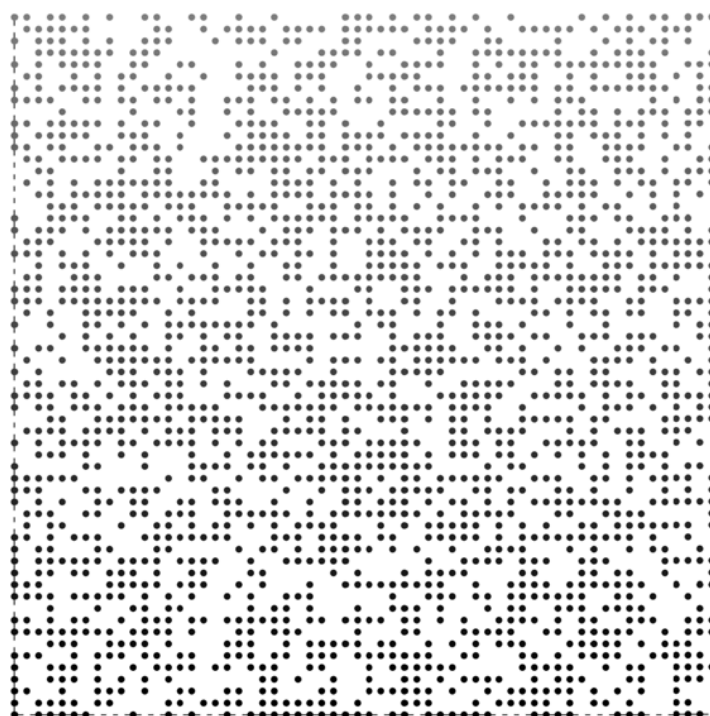
The reactive **current** through  $i$  and  $j$ :

$$F_{ij}^R = \pi_{ij}^R - \pi_{ji}^R = \rho_i t_{ij} (q_j - q_i)$$

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## Maze example

courtesy of Eric  
Vanden-Eijnden 2013



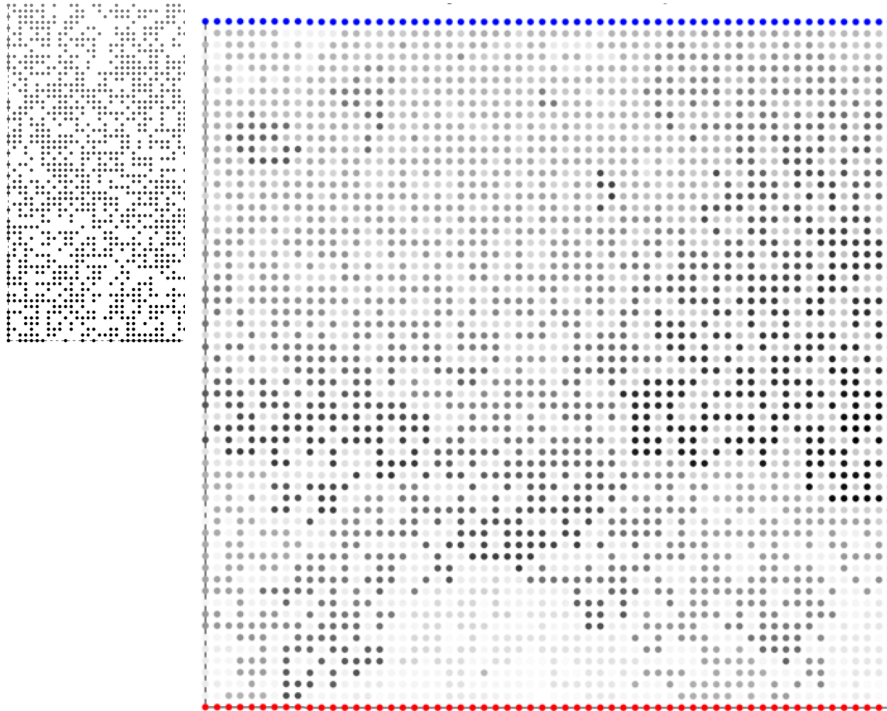
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# Maze example

courtesy of Eric  
Vanden-Eijnden 2013

Energy



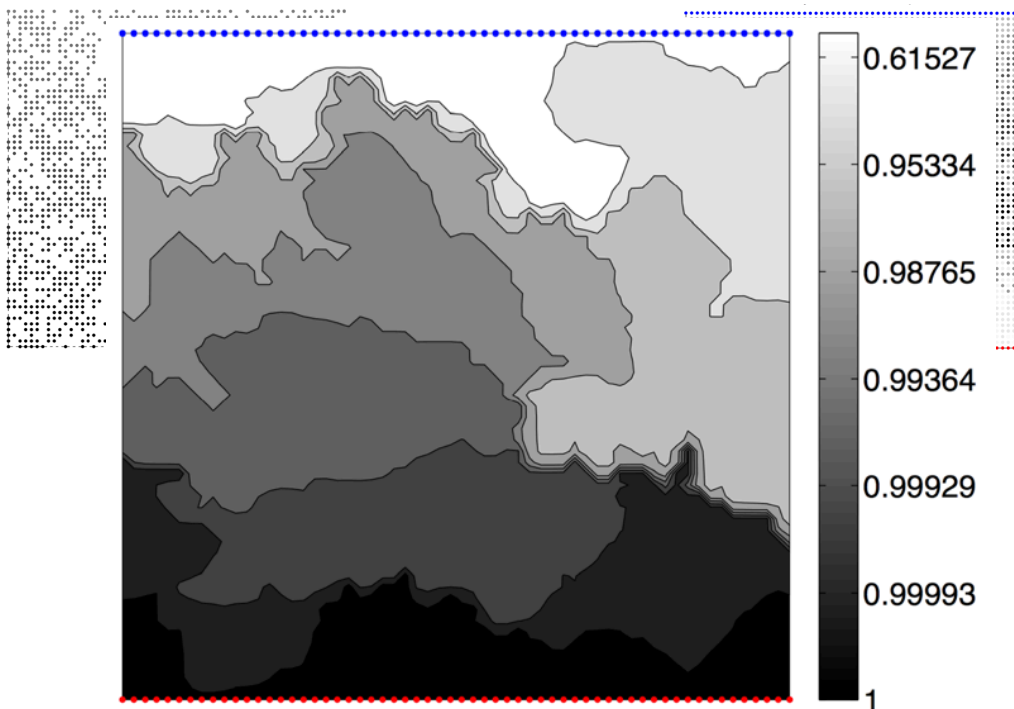
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# Maze example

courtesy of Eric  
Vanden-Eijnden 2013

Energy

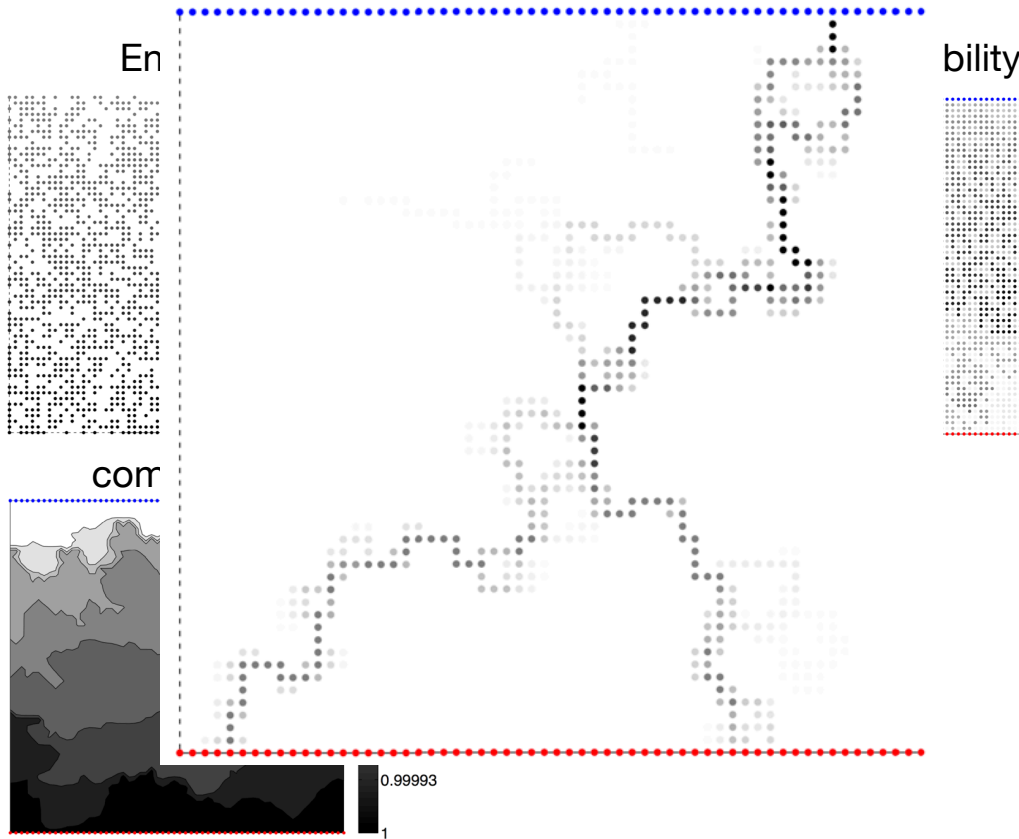
Reactive probability



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# Maze example

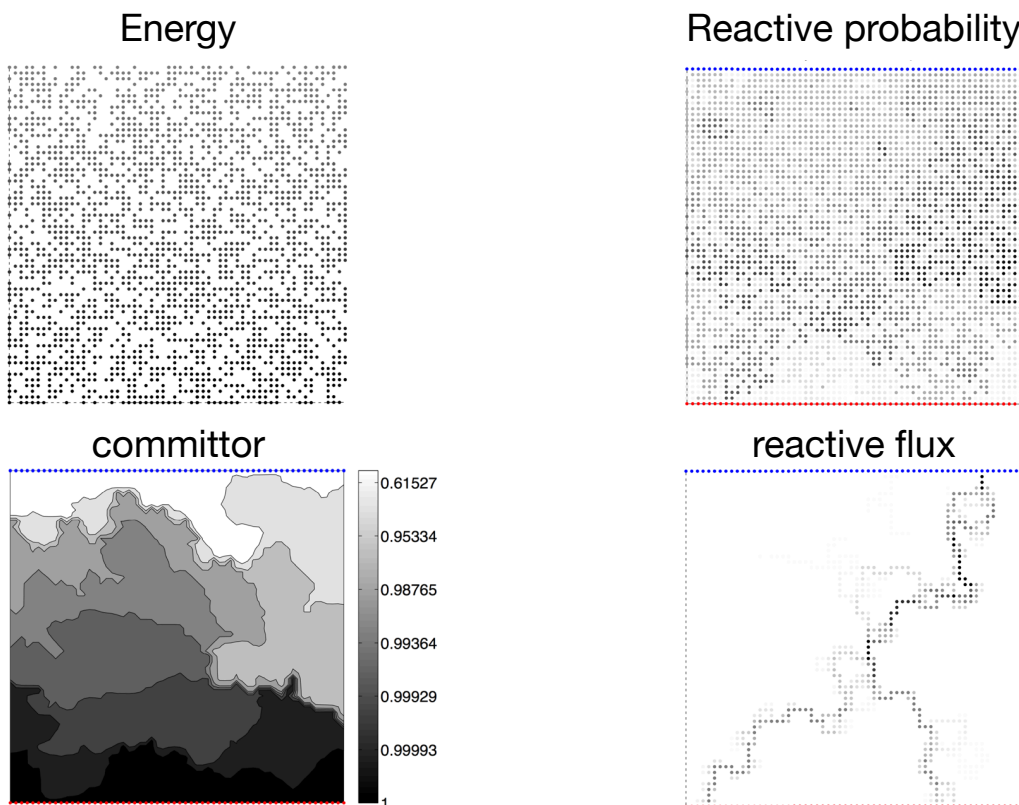
courtesy of Eric Vanden-Eijnden 2013



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# Maze example

courtesy of Eric Vanden-Eijnden 2013



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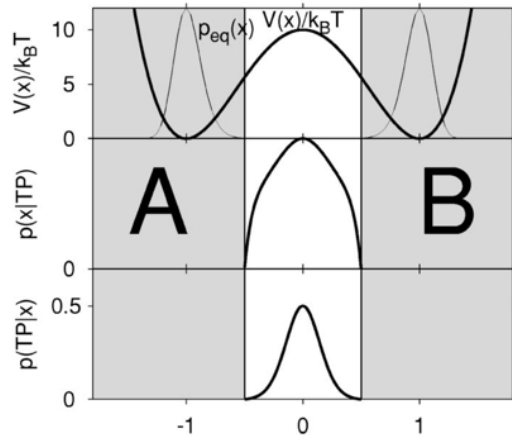


Gerhard Hummer

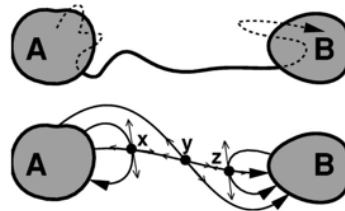
## More concepts

Even better definition for a transition state?

A transition state is defined as the state for which the probability that the state is crossed by a reactive trajectory is maximal.



$$p(\text{TP}|x) = 2q_A(x)q_B(x)$$



From transition paths to transition states and rate coefficients  
Gerhard Hummer, J. Chem. Phys. 120, 8 (2004)

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## Summary

### part 3

Transition Path Theory is a framework focused on sampling rare events in molecular simulation

- committor function, probability of states to reach product states, defines transition states and reaction coordinate
- reactive current/flux

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# Content

- Lecture 1: Markov processes
- Lecture 2: State clustering
- Lecture 3: Transition Path Theory
- **Lecture 4: Sampling Reactive Events**

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## Sampling Reactive Events

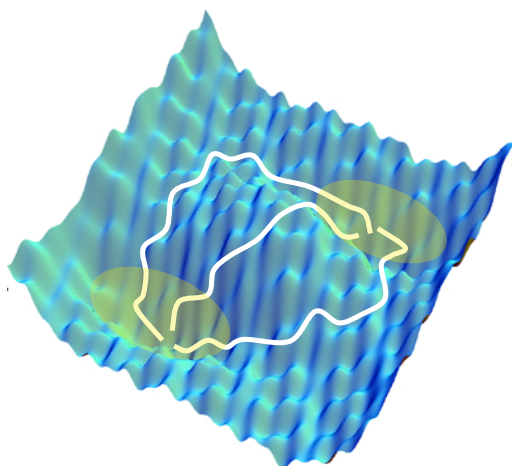
- **Generating reactive trajectories**
  - Transition Path Sampling, Transition Interface sampling, Forward Flux Sampling, Milestoning,...
- **Minimum (free) energy path**
  - String method, NEB, dimer method, minimum action
- **Free energy methods**
  - Umbrella Sampling, Jarzynski (steered) MD, constrained MD, adaptive force bias, metadynamics, path-metadynamics
- **Other**
  - Parallel tempering, Temperature-accelerated dynamics,...

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# Transition Path Sampling

Importance sampling of the path ensemble:  
all trajectories that lead over barrier and connect  
stable states.

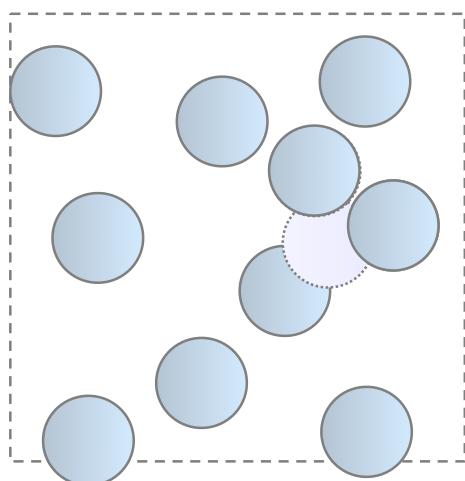
*P.G. Bolhuis, D. Chandler,  
C. Dellago, P.L. Geissler  
Annu. Rev. Phys. Chem 2002*



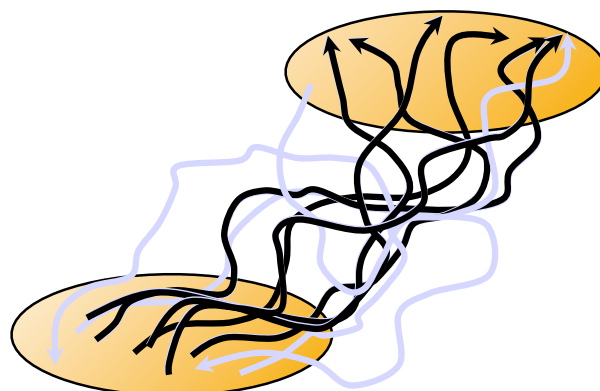
- No reaction coordinate (only state definition)
- True dynamics
- Works well for 2-state problems
- Needs an initial pathway
- Analysis required
- Transition state ensemble

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# Transition Path Sampling



Monte Carlo simulation:  
trials by random moves



Transition path sampling:  
trial paths by shooting move

TPS is Monte Carlo sampling of dynamical trajectories

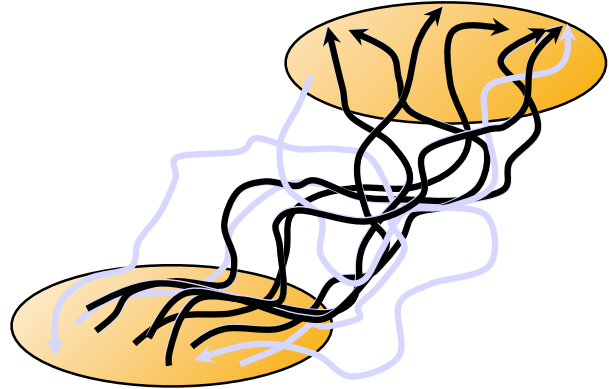
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# Transition Path Sampling

## Algorithm

1. generate an initial reactive path
2. choose a random configuration
3. make a small perturbation
4. integrate backward and forward
5. accept new path if it connects A and B, otherwise reject and go back to previous path
6. go back to 2.

In case of flexible path lengths, adjust acceptance criterion to maintain detailed balance.



Transition path sampling:  
trial paths by shooting move

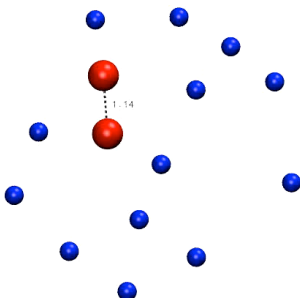
TPS is Monte Carlo sampling of dynamical trajectories

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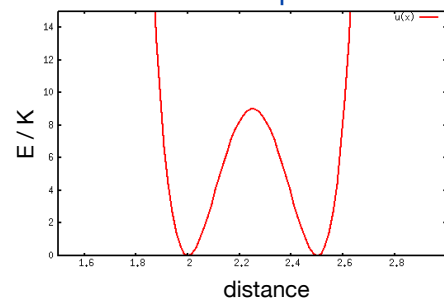
*Example: exercise  
from Molsim2014  
tutorial in Amsterdam*

## Transition Path Sampling

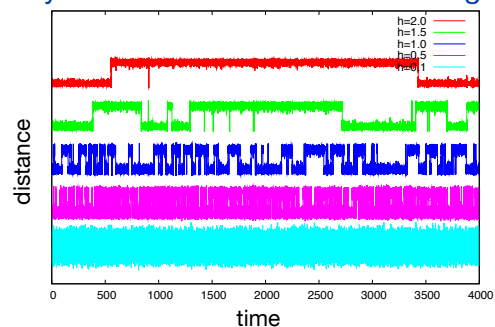
- 15 WCA particles in 2D box
- particles 1 and 2 are connected by double-well potential
- initial path from high-temperature MD simulation



double well potential

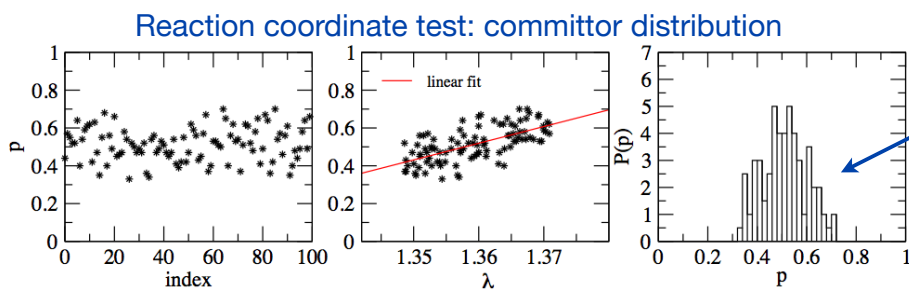
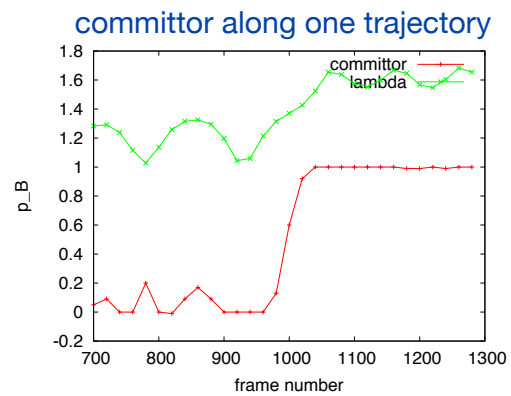
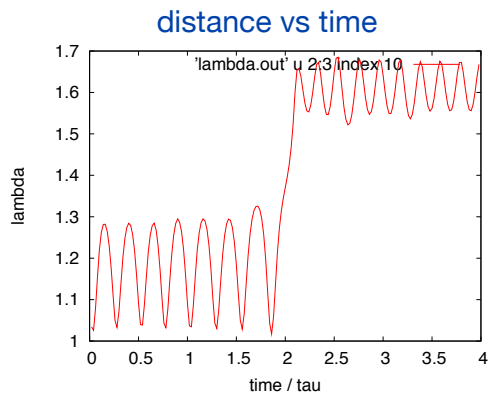


dynamics for different barrier heights



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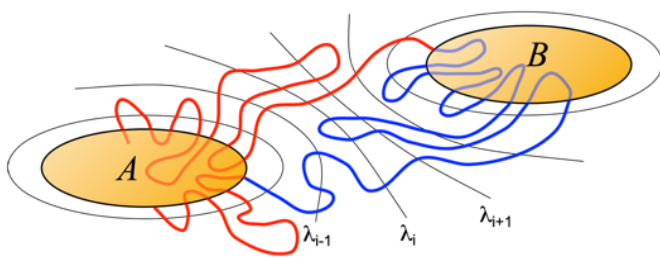
# Transition Path Sampling



Peaked committor probability means that the reaction coordinate is good

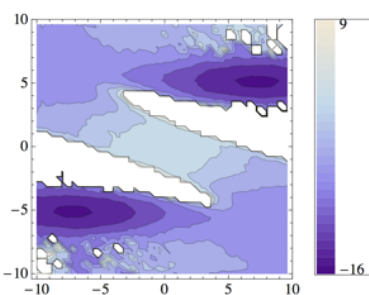
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# Transition Interface Sampling

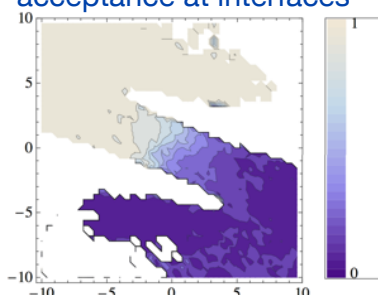


TPS between intermediate interfaces to sample "long" diffusive processes (protein folding) – no strong double-well behavior.

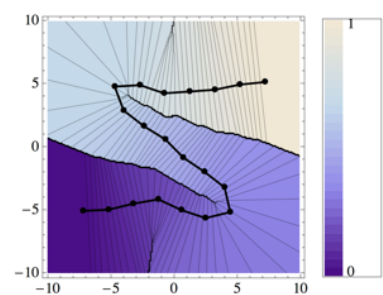
Difficult test: the Z-potential



committor from acceptance at interfaces

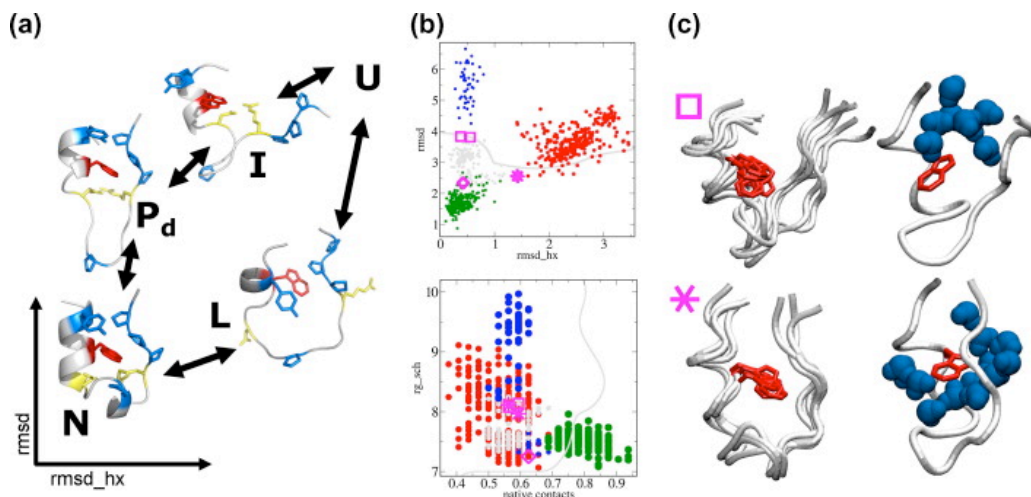


most likely pathway



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# Protein folding with TPS/TIS



(a) Schematic depiction of the folding mechanism of Trp-cage, elucidated with TPS simulations. (b) Stable states and transition states obtained by committor calculation, plotted as rgsch versus  $p$  and rmsd versus rmsdhx. Transition states for the NL and NPd transitions are plotted in magenta as stars and squares respectively. The scatter points are taken from the corresponding TPS trajectories. Native states (N) are plotted in green, loop structures (L) in red, close to native structures with Pro-12 detached (Pd) in gray and the I-state in blue. The region on the bottom of the gray line on the rmsd/rmsdhx (to the right on the  $p$ /rgsch) plot is the part of the configuration space not reachable by our REMD simulation starting from an unfolded configuration. (c) Left: superimposed TS structures for four different NL (star) and six for PdI paths (square). Right: one of the TS structures for both routes in a side view plotted as a ribbon. Central tryptophan residue is plotted in licorice representation (red), water molecules within 5 Å of Trp-6 in space-filling representation in blue. (For interpretation of the references to colour in this figure legend, the reader is referred to the web version of this article.)

Transition path sampling of protein conformational changes  
 J. Juraszek, J. Vreeede and P.G. Bolhuis  
*Chem. Phys.* **396** (2012), 30-44

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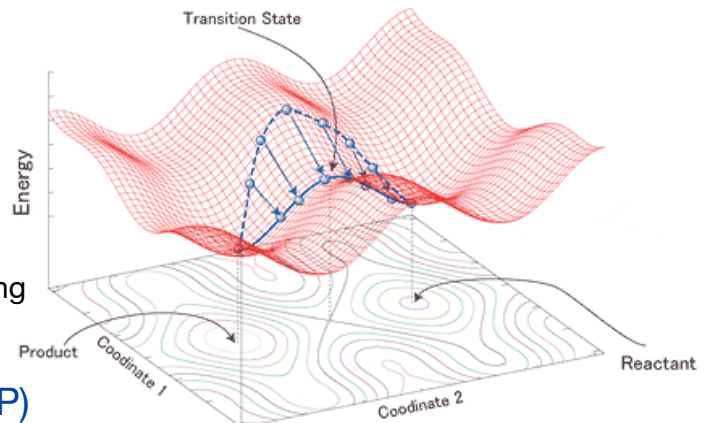
## Path finding

### Minimum Energy Path (MEP)

- string method, Nudge Elastic Band, ...
- start from guess path
- string of equidistant nodes from A to B
- compute gradients at nodes
- move nodes steepest descent
- maintain equidistant nodes
- converged: forces perpendicular to string are zero

### Minimum Free Energy Path (MFEP)

- finite temperature string method
- perform constrained MD simulation at each node
- compute average force of constraint
- steepest descent perpendicular to string
- converged: perpendicular forces are zero
- free energy profile estimated by integration of constraint force along the string from A to B



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# Exploring the free energy landscape

*Escaping free-energy minima,*

Laio and Parrinello, PNAS (2002)

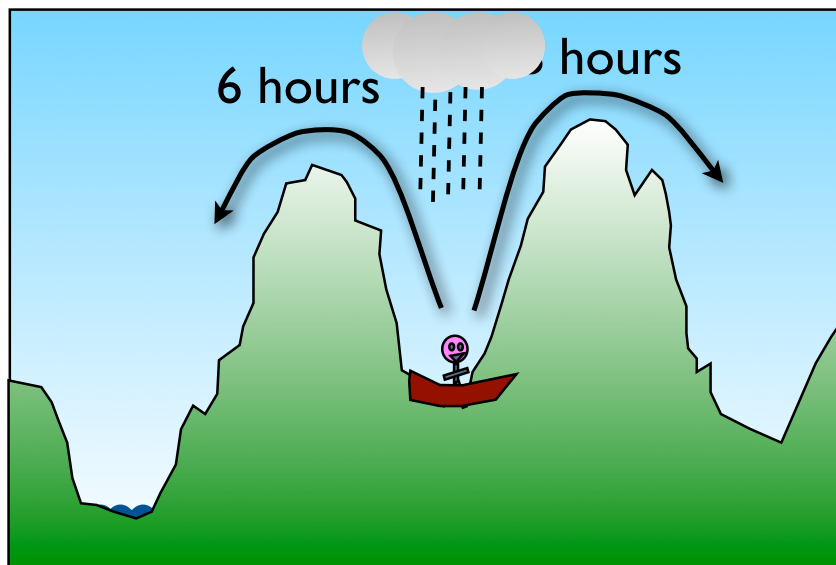
$$V(t, s) = \sum_{t' < t} H_{t'} \prod_{\alpha} \exp \left[ \frac{-(s_{\alpha} - s_{\alpha}^{t'})^2}{2\delta_{\alpha}^2 W^2} \right]$$

- enhance sampling
- probe free energy landscape
- multiple collective variables

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## Metadynamics

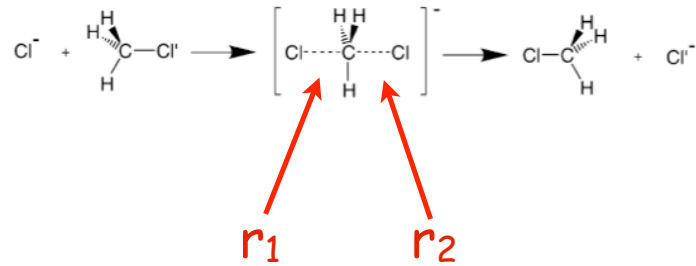
Metadynamics, Laio and Parrinello, PNAS (2002)



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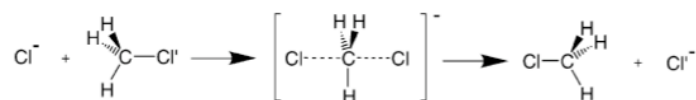
## simple example: $S_N2$ reaction



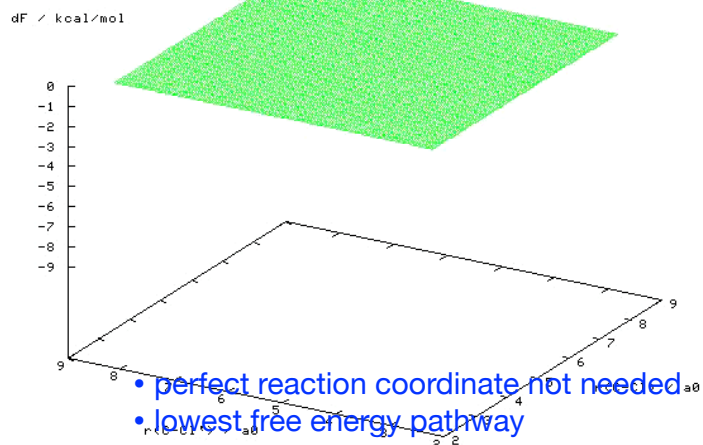
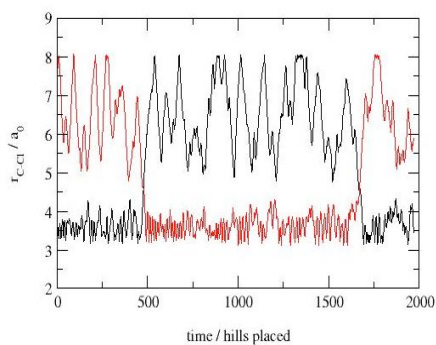
Two collective variables:  $r_1$  and  $r_2$

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## simple example: $S_N2$ reaction



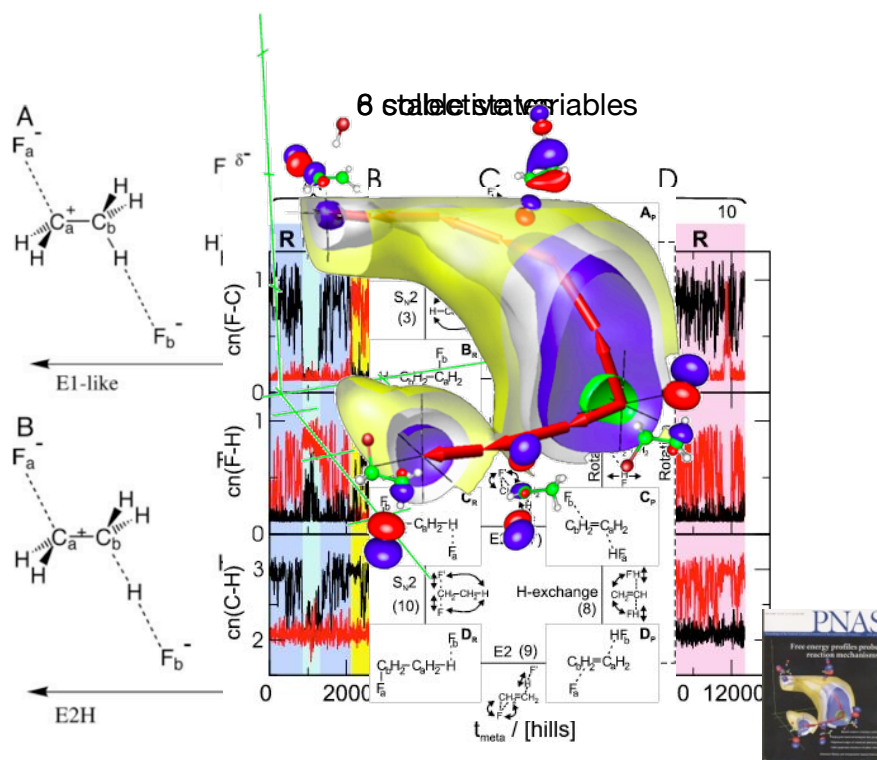
Two collective variables:  $r_1$  and  $r_2$



Bernd Ensing, Alessandro Laio, Michele Parrinello and Michael L. Klein  
*J. Phys. Chem. B* **109** (2005), 6676-6687

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# *E2 vs S<sub>N</sub>2 reaction*



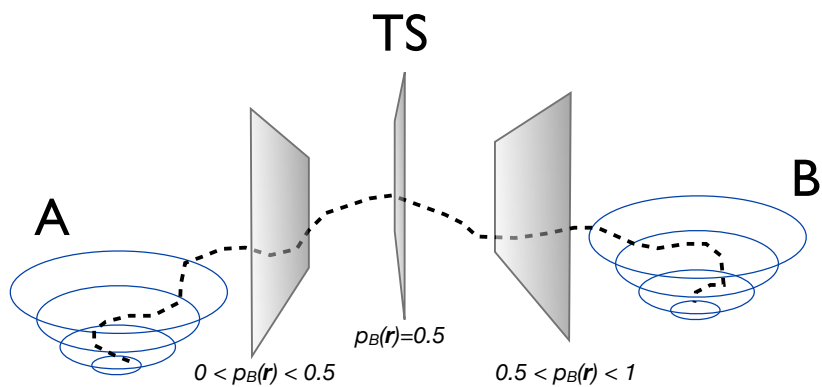
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*More collective variables?*

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# *transition path sampling and path-metadynamics*

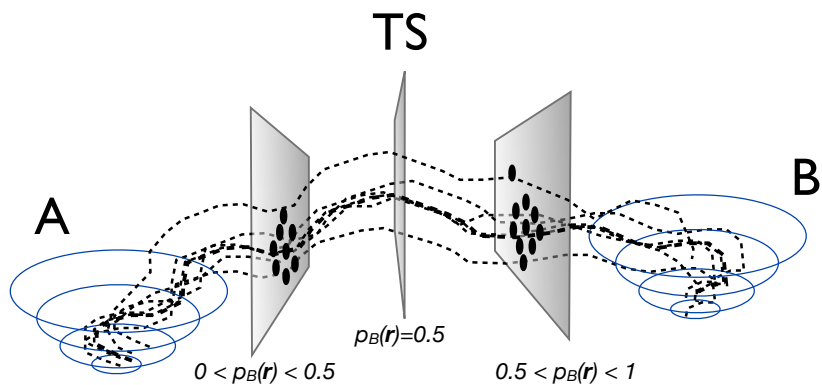
iso-committor surfaces



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# *transition path sampling and path-metadynamics*

iso-committor surfaces

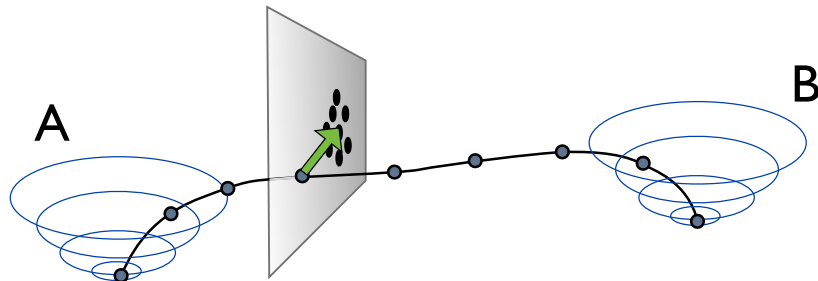


the average transition pathway  
(in CV space)

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# *path-metadynamics*

distance to mean density



- start from guess path
- bias dynamics along path
- move nodes to the mean density
- maintain equidistant nodes

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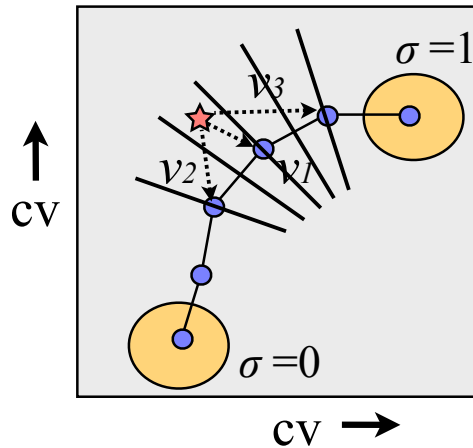
## *MFEP as the collective variable*

### Metadynamics

- add extra collective variable:  $\sigma$
- $\sigma$  is a function of all other coll. variables
- biasing potential is only working on  $\sigma$
- $\sigma$  function adapts on the fly

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## *MFEP as the collective variable*

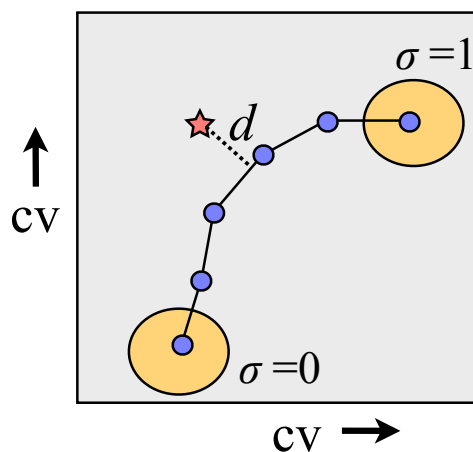


MFEP CV

$$\sigma = n^{-1} \left( i_{min} \pm \frac{\sqrt{(\bar{v}_1 \bar{v}_3)^2 - \bar{v}_3 \bar{v}_3 (\bar{v}_1 \bar{v}_1 - \bar{v}_2 \bar{v}_2)} - \bar{v}_1 \bar{v}_3}{2(\bar{v}_3 \bar{v}_3 - 1)} \right)$$

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## *MFEP as the collective variable*



histogram distance to the path for each node  
 maximum of  $P(d)$  should be at  $d=0$

or accumulate average  $d$  for each node

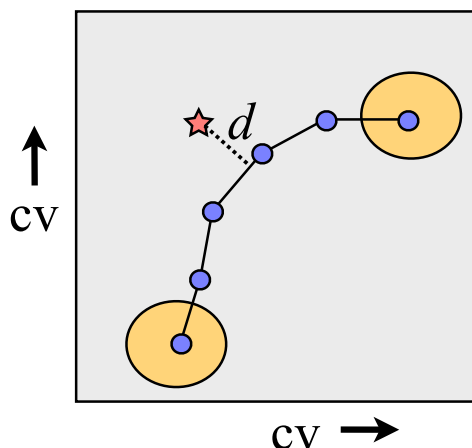
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# MFEP as the collective variable

$$\langle d_i \rangle = \frac{\sum_0^\tau w_i * d_i}{\sum_0^\tau w_i}$$

$$w_i(t) = (\sigma(n-1) - i_{min}) f^{\tau-t}$$

$$f = \exp[\tau^{-1} * \ln \frac{1}{2}]$$

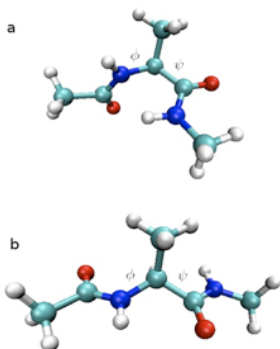


move nodes to  $d$  (every step)  
set  $d=0$   
redistribute nodes along path

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• Zhiwei Liu  
• Preston Moore  
University of the Sciences, Philadelphia

## alanine dipeptide



Gas-phase

**Classical Molecular Dynamics**  
(CM3D code by Preston Moore, USP)

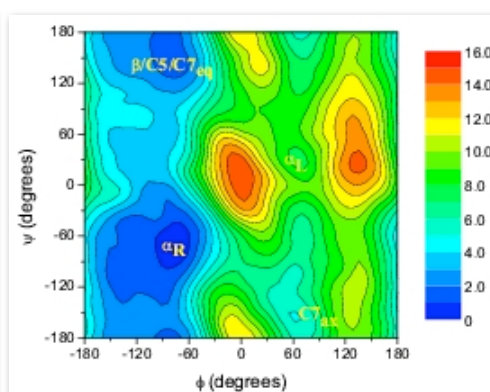
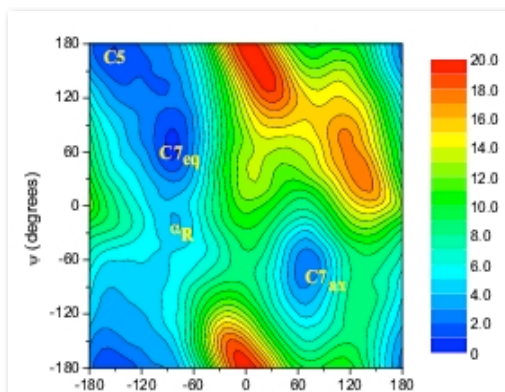
Alanine dipeptide + 216 water  
CHARMM27 forcefield (modified internal params water)  
Cubic periodic box L=18.8 Angstrom  
NVT ensemble, T=298K

Hill size: W=0.2 rad, H=0.02 kcal/mol  
Hill stride: 100 MD steps (=50 fs)

**Total simulation time: ca. 5 ns**

**Error in energy: 0.3-0.5 kcal/mol**  
**Error in angles: 3-10 degrees**

Aqueous solution



Metadynamics as a tool for exploring the free energy landscape of chemical reactions.  
Bernd Ensing, Marco De Vivo, Zhiwei Liu, Preston Moore, and Michael L. Klein  
Acc. Chem. Res. **39** (2006), 73-81

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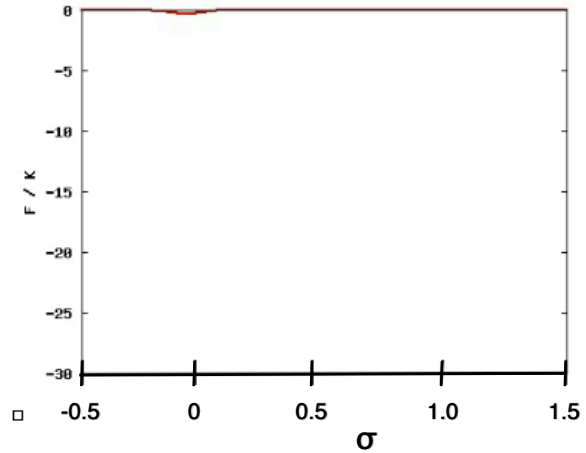
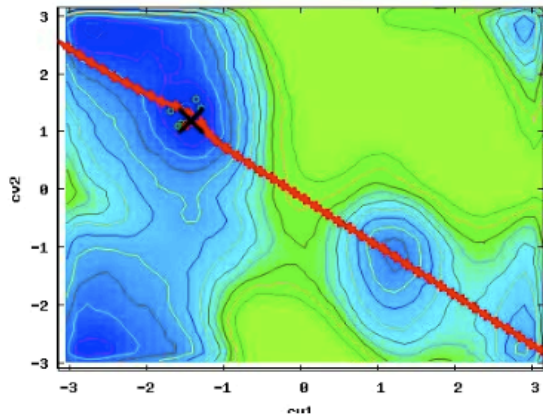
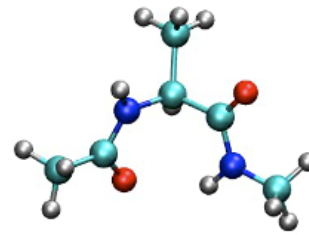
# alanine dipeptide

## Parameters:

$T = 300$  K  
 $H_{\text{gaussian}} = 10$  K  
 $W_{\text{gaussian}} = 0.05$   
 $\Delta t_{\text{gaussian}} = 100$  MD steps  
 $n = 20 + 20 + 20$   
 $\tau = 1000$  MD steps

## every recrossing:

H x 50%  
W x 50%  
 $\tau \times 100$



Path Finding on High-Dimensional Free Energy Landscapes.  
Grisell Díaz Leines and Bernd Ensing  
*Phys. Rev. Lett.* **109** (2012), 020601

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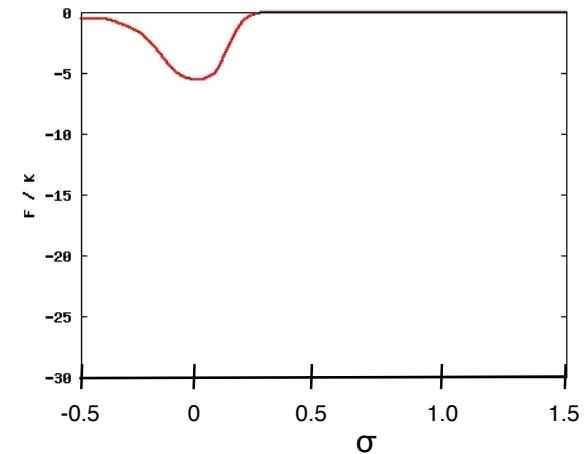
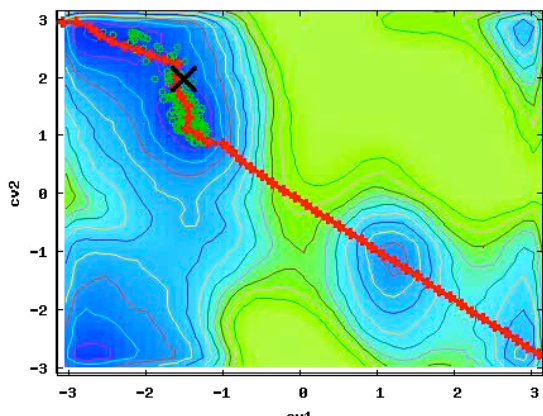
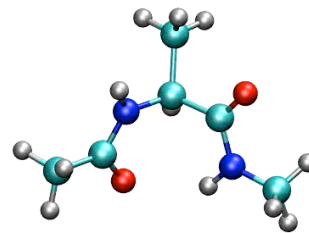
# alanine dipeptide

## Parameters:

$T = 300$  K  
 $H_{\text{gaussian}} = 10$  K  
 $W_{\text{gaussian}} = 0.05$   
 $\Delta t_{\text{gaussian}} = 100$  MD steps  
 $n = 20 + 20 + 20$   
 $\tau = 1000$  MD steps

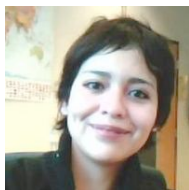
## every recrossing:

H x 50%  
W x 50%  
 $\tau \times 100$



Path Finding on High-Dimensional Free Energy Landscapes.  
Grisell Díaz Leines and Bernd Ensing  
*Phys. Rev. Lett.* **109** (2012), 020601

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Grisell Diaz Leines

# path-metadynamics

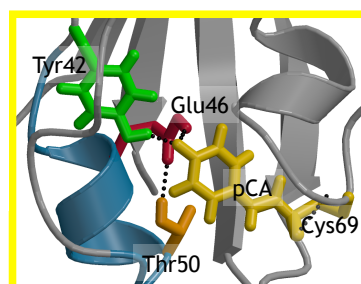
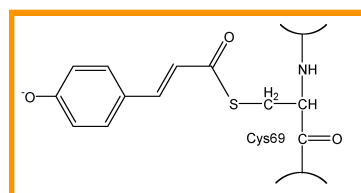
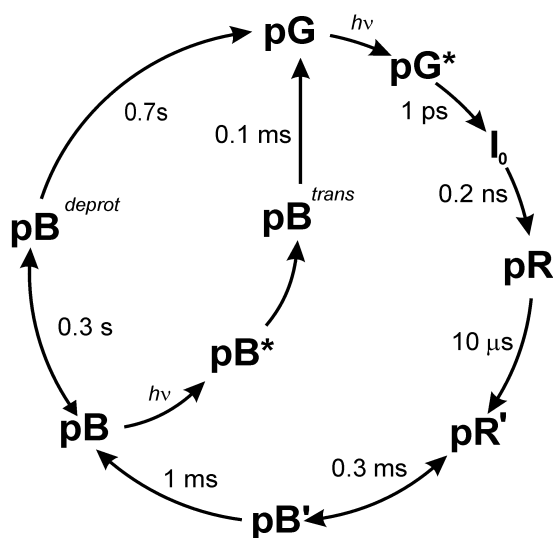


Davide Branduardi  
MPI-Biophys, Frankfurt

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## Photoactive Yellow Protein

blue-light sensor from *H. halophila*

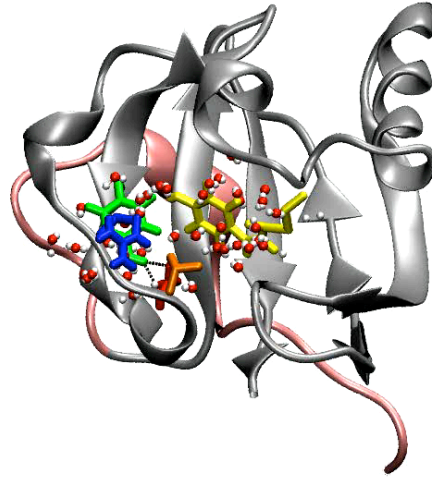
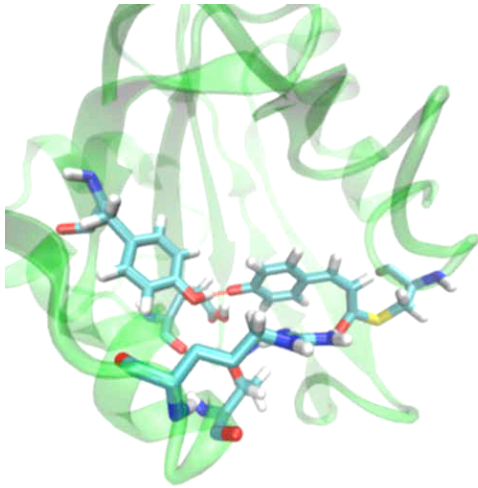


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# Photoactive Yellow Protein

Proton transfer step

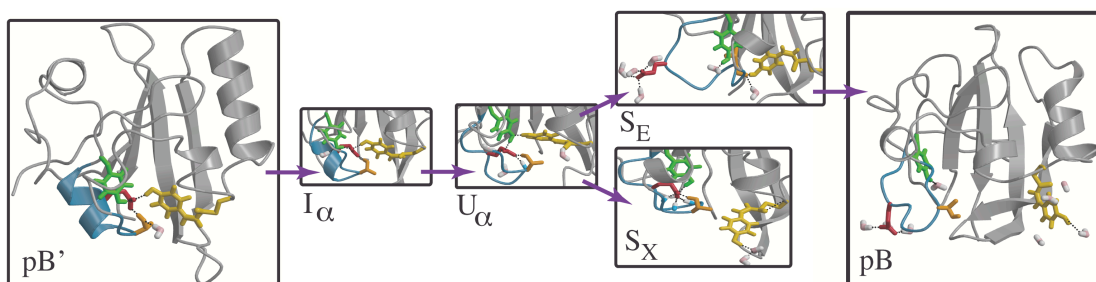
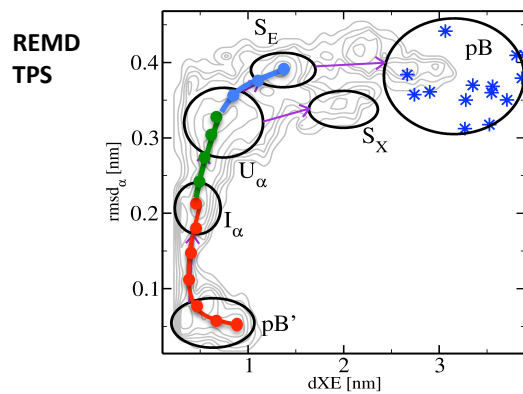
Conformational transformation



Predicting the signaling state of Photoactive Yellow Protein.  
J. Vreede, W. Crielaard, K.J. Hellingwerf, and P.G. Bolhuis  
*Biophys J.* **88** (2005), 3525 DOI: [10.1529/biophysj.104.055103](https://doi.org/10.1529/biophysj.104.055103)

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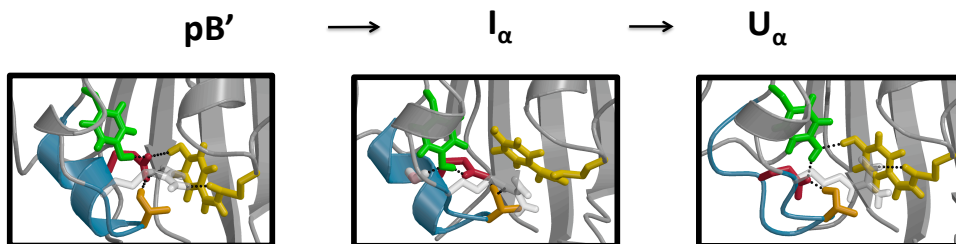
## transition path sampling



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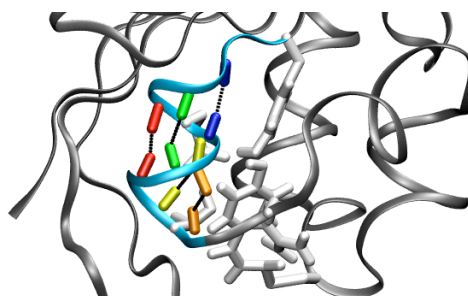
# path-metadynamics study of PYP

unfolding of helix 43-51

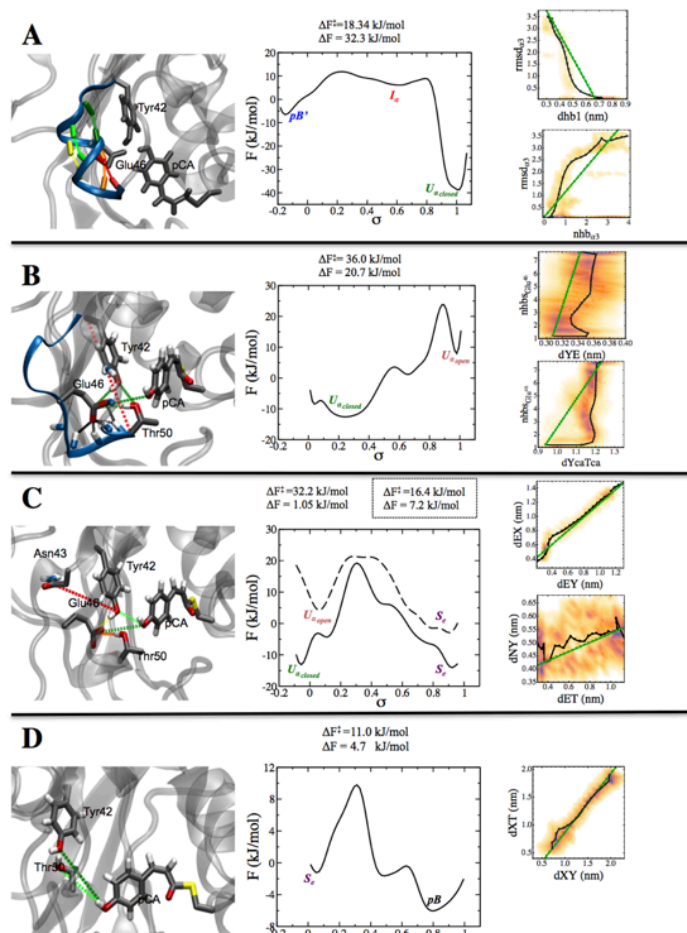


## Collective variables:

- Asn43-O---Gly47-H =  $dhb_{\alpha_3}^1$
- Ala44-O----Asp48-H= $dhb_{\alpha_3}^2$
- Ala45-O----Ile49-H=  $dhb_{\alpha_3}^3$
- Glu46-O----Thr50-H= $dhb_{\alpha_3}^4$
- Gly47-O----Gly51-H= $dhb_{\alpha_3}^5$



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# Content

- Lecture 1: Markov processes
- Lecture 2: State clustering
- Lecture 3: Transition Path Theory
- Lecture 4: Sampling Reactive Events

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## Summary

### part 1

A stochastic hopping between states is Markovian if the probability to reach the next state does not depend on the past (no memory)

- reversible Markov chains (detailed balance)
- ergodicity
- spectral decomposition (convergence to equilibrium)

### part 2

Markov State Modeling applied as analysis tool of the dynamics of a small peptide

- Analysis of molecular dynamics simulation
- Clustering based on structures
- Relation to free energy landscape based on cluster size
- Transition probability between clusters (states)
- Using the MSM the dynamics can be sampled in a coarse-grain manner, e.g. using kinetic Monte-Carlo

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# Summary

## part 3

Transition Path Theory is a framework focused on sampling rare events in molecular simulation

- committor function, probability of states to reach product states, defines transition states and reaction coordinate
- reactive current/flux

## part 4

Advanced sampling methods for rare events focus on sampling reactive trajectories, the free energy landscape, or an average transition pathway

- Transition path sampling (TPS/TIS/FFS)
- String method
- Metadynamics, path-metadynamics

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## Further Reading

- **The life and work of A.A. Markov**, Basharin, G.P., Langville, A.N., Naumov, V.A., Linear Algebra and its Applications 386 (2004), 3-26.
- **The activated complex and the absolute rate of chemical reactions.** Eyring H. Chem. Rev. 17 (1935), 65–77
- **The transition state method.** Wigner E. Trans. Faraday Soc. 34 (1938), 29–41
- **Understanding Molecular Simulation: From Algorithms to Applications.** Frenkel, D.; Smit, B., Academic, San Diego, CA, 2 edition, 2002.
- **Transition-Path Theory and Path-Finding Algorithms for the Study of Rare Events** Weinan E and Eric Vanden-Eijnden Annu. Rev. Phys. Chem. 61 (2010), 391–420
- **String method for the study of rare events.** E W, Ren W, Vanden-Eijnden E. Phys.Rev.B 66 (2002), 052301
- **Finite temperature string method for the study of rare events.** E W, Ren W, Vanden-Eijnden E. J. Phys. Chem. B 109 (2005), 6688–93
- **Reactive flux and folding pathways in network models of coarse-grained protein dynamics.** Alexander Berezhkovskii, Gerhard Hummer, and Attila Szabo J. Chem. Phys. 130 (2009), 205102
- **A growing string method for determining transition states: Comparison to the nudged elastic band and string methods.** Peters, B.; Heyden, A.; Bell A. T.; Chakraborty A. J. Chem. Phys. 2004, 120, 7877–7886.
- **Transition path sampling and the calculation of rate constants.** Dellago, C.; Bolhuis, P. G.; Csajka, F. S.; Chandler, D.. J. Chem. Phys. 108 (1998), 1964–1977.
- **Non-linear reaction coordinate analysis in the reweighted path ensemble.** Wolfgang Lechner, Jutta Rogal, Jarek Juraszek, Bernd Ensing, and Peter Bolhuis J. Chem. Phys. 133 (2010), 174110
- **Non-linear reaction coordinate analysis in the reweighted path ensemble.** Wolfgang Lechner, Jutta Rogal, Jarek Juraszek, Bernd Ensing, and Peter Bolhuis J. Chem. Phys. 133 (2010), 174110
- **From transition paths to transition states and rate coefficients** Gerhard Hummer, J. Chem. Phys. 120 (2004), 8
- **The reweighted path ensemble.** Jutta Rogal, Wolfgang Lechner, Jarek Juraszek, Bernd Ensing, and Peter Bolhuis J. Chem. Phys. 133 (2010), 174109
- **Nudged elastic band method for finding minimum energy paths of transitions.** In Classical and Quantum Dynamics in Condensed Phase Simulations, Jonsson H, Mills G, Jacobsen K W.1998. ed. B J Berne, G Ciccotti, D F Coker, pp. 385–404. Singapore: World Sci.
- **A dimer method for finding saddle points on high dimensional potential surfaces using only first derivatives.** Henkelman G, Jonsson H. J. Chem. Phys. 111 (1999), 7010–22
- **Conjugate peak refinement: an algorithm for finding reaction paths and accurate transition states in systems with many degrees of freedom.** Fischer S, Karplus M. Chem. Phys. Lett. 194 (1992), 252–61
- **A milestone study of the kinetics of allosteric transition: atomically detailed simulations of deoxy Scapharca hemoglobin.** Elber R. Biophys. J. 92 (2007), L85–87
- **Forward flux sampling-type schemes for simulating rare events: efficiency analysis.** Allen R J, Frenkel D, ten Wolde P R. J. Chem. Phys. 124 (2006.), 194111
- **Finding transition pathways using the string method with swarms of trajectories.** Pan AC, Sezer D, Roux B. J. Phys. Chem. B 112 (2008), 3432–40
- **Escaping free-energy minima.** Laio, A.; Parrinello, M. Proc. Natl. Acad. Sci. USA, 99 (2002), 12562–12566.
- **A recipe for the computation of the free energy barrier and the lowest free energy path of concerted reactions.** Ensing, B.; Laio, A.; Parrinello, M.; Klein, M. L. J. Phys. Chem. B 109 (2005), 6676–6687
- **Path Finding on High-Dimensional Free Energy Landscapes.** Grisell Diaz Leines and Bernd Ensing Phys. Rev. Lett. 109 (2012), 020601

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